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# SYNTHESIS AND CHARACTERISATION OF NEW METALLACYCLES CONTAINING $[Ph_2P(E)NP(E)Ph_2]^-$ (E = S OR Se) LIGANDS

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<u>Abstract</u> Reaction of [Ph<sub>2</sub>P(E)NP(E)Ph<sub>2</sub>] (E = S or Se) with a series of late transition-metal dimers, in the or MeOH, leads to facile bridge cleavage and formation of new mononuclear compounds.

# **INTRODUCTION**

The monoanions [Ph<sub>2</sub>P(E)NP(E)Ph<sub>2</sub>] (E = O, S or Se) are excellent ligands for metal ion complexation. One particularly attractive feature of [Ph<sub>2</sub>P(E)NP(E)Ph<sub>2</sub>] is their ready synthesis from commercially available (Ph<sub>2</sub>P)<sub>2</sub>NH and the appropriate chalcogen followed by deprotonation with base (e.g. KOBu<sup>t</sup>). Several recent reports have described new complexes of [Ph<sub>2</sub>P(E)NP(E)Ph<sub>2</sub>] (E = S or Se) with Group VIII metals such as Rh(I), Pd(II) and Pt(II).<sup>2-4</sup> Here we present some new examples of six-membered ME<sub>2</sub>P<sub>2</sub>N (E = S or Se) metallacycles with Ru(II), Rh(III), Pd(II) and Pt(II).<sup>5</sup> In these complexes, [Ph<sub>2</sub>P(E)NP(E)Ph<sub>2</sub>] chelates to the metal centre in a bonding mode typical for this class of ligand.

### RESULTS AND DISCUSSION

The potassium salts  $K[Ph_2P(E)NP(E)Ph_2]$  (E = S I; E = Se II) were synthesised according to known methods.<sup>3,6</sup> Previous work has shown that the chloro bridges in [{MCl( $\mu$ -Cl)(PMe<sub>2</sub>Ph)}<sub>2</sub>] (M = Pd, Pt) or

[{Rh( $\mu$ -Cl)(cod)}<sub>2</sub>] (cod = cycloocta-1,5-diene) are cleaved with II to give [MCl{Ph<sub>2</sub>P(Se)NP(Se)Ph<sub>2</sub>-Se,Se'}PMe<sub>2</sub>Ph] or [Rh{Ph<sub>2</sub>P(Se)N-P(Se)Ph<sub>2</sub>-Se,Se'}cod] respectively.<sup>3</sup> In a similar manner the new neutral complexes 1 - 12 were prepared as depicted in Scheme 1. All compounds were isolated in good to excellent yields (64 - 97%), are air stable in the solid state and in solution and display the expected analytical and spectroscopic data. The <sup>31</sup>P{<sup>1</sup>H} NMR data (CDCl<sub>3</sub>) indicate single phosphorus containing species present in solution with  $\delta$ (P) for the sulfur analogues ca. 10 ppm to higher frequency than their selenium analogues. Furthermore <sup>1</sup>J(PSe) (typically 500 - 600 Hz) is reduced in magnitude relative to the free ligand II [687 Hz, (CD<sub>3</sub>)<sub>2</sub>SO].<sup>3</sup> The X-ray crystal structure of a representative example, [Pd(C<sub>9</sub>H<sub>12</sub>N){Ph<sub>2</sub>P-(Se)NP(Se)Ph<sub>2</sub>-Se,Se'}] 4, is shown in Figure 1.

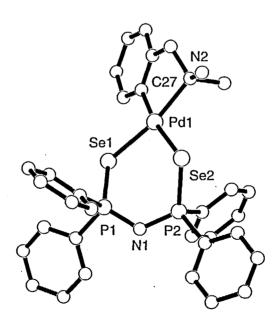
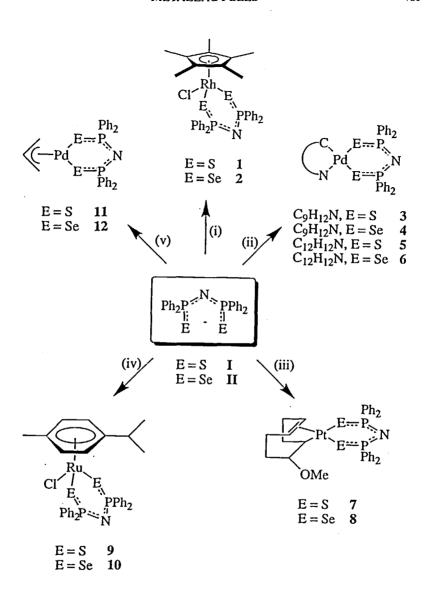


FIGURE 1 X-ray crystal structure of 4



In conclusion we have shown that reaction of I (or II) with several transition-metal reagents affords new M-E-P-N-P-E metallacycles (E = S or Se). The co-ordinated ligand adopts a familiar chelating mode of bonding. Full details of syntheses / characterisation will be reported elsewhere.

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